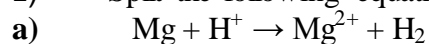


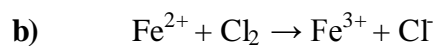
REDUCTION/OXIDATION EQUATION EXERCISES

1) Split the following equations into **half equations**.



Reduction

Oxidation



Reduction

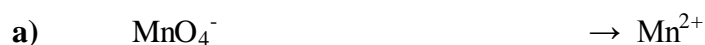
Oxidation



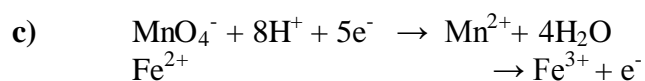
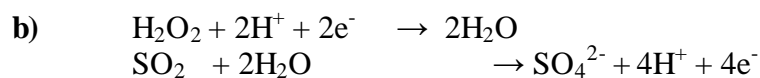
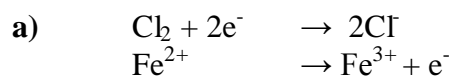
Reduction

Oxidation

2) **Balance** the following **half equations**.



3) **Combine** the following **half equations**.



4) Write balanced half equations for each oxidising agent, state any observations



5) Write a balanced half equation for each reducing agent, state any observations

