

# Isomers

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The atoms for any given \_\_\_\_\_ formula can be arranged in different ways. The different structures that can be made from the molecular formula are called \_\_\_\_\_.

The structural isomers of  $C_4H_{10}$  are



1. Draw the structural isomers of  $C_5H_{12}$

2. Draw the structural isomers of  $C_6H_{14}$

# Condensed Structural Formula

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The \_\_\_\_\_ represents how the atoms in an organic compound are joined to each other.

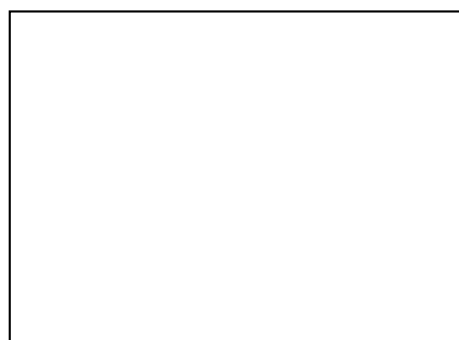
The structural formulas can be drawn in a \_\_\_\_\_ form

Each carbon atom in the longest chain is written with atoms attached to it to the right of it.

## Structural formula

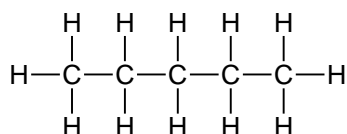


## Condensed Structural formula

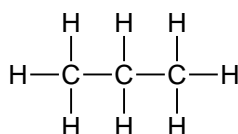


3. Draw the condensed structural formula for the following molecules

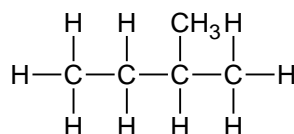
a)



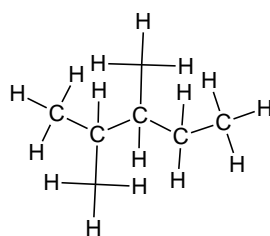
b)



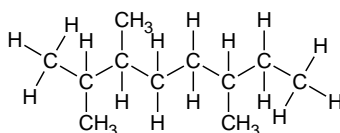
c)



d)



e)



f)

